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The Mollier diagram,  
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also called the enthalpy (h) – entropy (s) chart or h-s chart, is a graphical representation of thermodynamic properties of materials. In general, it is a relationship between enthalpy (measure of the energy of a thermodynamic system), air temperature, and moisture content.

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expansion process  
through an adiabatic  
turbine. Line 1–2

represents the actual  
expansion and line 1–2<sub>s</sub>  
the ideal or reversible  
expansion. The fluid  
velocities at entry to and  
exit from a turbine may  
be quite high and the  
corresponding kinetic  
energies significant.

**Mollier Diagram - an**

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Engineering the H–S

chart or Mollier

diagram, plots the total

heat against entropy,

describing the enthalpy

of a thermodynamic

system. A typical chart

covers a pressure range

of 0.01–1000 bar, and

temperatures up to 800

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degrees Celsius.

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design tool for building engineers and designers. common psychrometrics terms; The Mollier diagram is a variant of the psychrometric chart.

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For Thermal Engineering

diagram, plots the total heat against entropy, describing the enthalpy of a thermodynamic system. A typical chart covers a pressure range of 0.01–1000 bar, and temperatures up to 800 degrees Celsius. It shows enthalpy  $H$  in terms of internal energy  $U$ , pressure  $p$

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$p$  and volume  $V$  using the relationship  $H = U + pV$ .

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provide water (liquid) properties. A sample of the diagram is shown below in order to illustrate the main points of the diagram and how to use the diagram.

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