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Buffer solutions are a mixture of substances that have a fairly constant pH regardless of addition of acid or base. They are used in

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medicine, industry and manufacturing to keep the pH of a solution stable regard- less of other reactions that might be occurring.

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POGIL 20. One buffer system in the bloodstream of animals is the carbonic acid/bicarbonate buffer. $\text{H}_2\text{CO}_3 + \text{H}_2\text{O} \rightleftharpoons \text{HCO}_3^- + \text{H}_3\text{O}^+$ The carbonic acid in the bloodstream is formed by the combination of carbon ...

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POGIL Activities for AP* Chemistry-modified Name Buffers.

POGIL™ Activities for AP* Chemistry-modified 3 d) Now suppose one keeps adding 1.0 M HCl to this buffer until a total of 101 mL of 1.0M HCl have been added. At this point, the pH of the solution will be very acidic.

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