

### Section 2 Reinforcement Chemical Bonds Answers

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Section 2 Types of Bonds. Gain or loss of electrons. Atoms gain or lose electrons to become stable. Atoms that gain or lose electrons are called ions. It is the electric forces between oppositely charged particles, such as ions, that hold compounds together. Bond Formation.

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freezing point = 0&deg;C, boiling point = 100&deg;C 2. an upward Study Guide and Reinforcement 17 ANSWER KEY Chapter 17 Section 3 Section 1 1. atomic number 1. bubble chamber 2. element 2. particle accelerator 3. chemical symbol 3.

*Study Guide and Reinforcement - Answer Key*  
1. ionic bond. 2. ion. 3. chemical bond. 4. compound. 5. covalent bond. 6. molecule. 7. formula. 8. electron dot diagram. 9. polar bond. 10. polar molecule ... electrons are shared equally by atoms. In a polar bond, electrons are shared unequally. (3/2) Reinforcement. Section 1 (page 81) 1. the formation of new substances that have. properties ...

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2. 3. reaction force 4. action force 5. The force also will be 500 N because action-reaction forces are equal and opposite. 6. p = m v = 2 kg 10 m/s = 20 kg • m/s 7. p = m v = 2000 kg 10 m/s = 20,000 kg • m/s 8. the 2000-kg truck because it has a greater mass Chapter 4 1. energy 2. potential 3. kinetic 4. gravitational 5. speed Section 1 ...

*Study Guide and Reinforcement - Answer Key*  
Section 2 Reinforcement Classifying Chemical Reactions Answers Steel bars and wire are traditional types of reinforcement that have excellent high tensile yield properties, ductility and moduli of elasticity that complement concrete's attributes. To provide added durability when required, black steel bars are galvanized or epoxy-coated.

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Chemical Bonds Answer the questions about the diagram shown below. Write your answers in the spaces provided. 12P 12n How many electrons will atom A lose to atom B? 2. kind of chemical bond will formed between atom and atom B if atom A loses lectrons and atom B gains these electrons? 3.

*Quia*  
SECTION 2.4 CHEMICAL REACTIONS Reinforcement KEY CONCEPT Life depends on chemical reactions. At the most fundamental level, every process that takes place in an organism depends on chemical reactions. In a chemical reaction, substances are changed into different substances by the breaking and forming of chemical bonds. The substances that are

*SECTION CHEMICAL REACTIONS 2.4 Reinforcement*  
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